

6-Bromo-1-methyl-4-[2-(4-methylbenzylidene)hydrazinylidene]-3H-2λ⁶,1-benzothiazine-2,2-dione

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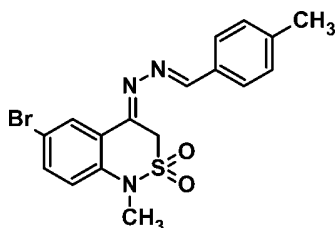
Received 11 July 2011; accepted 15 July 2011

Key indicators: single-crystal X-ray study; $T = 296$ K; mean $\sigma(\text{C}-\text{C}) = 0.004$ Å; R factor = 0.034; wR factor = 0.078; data-to-parameter ratio = 18.7.

In the title compound, $\text{C}_{17}\text{H}_{16}\text{BrN}_3\text{O}_2\text{S}$, the two fused rings are twisted by a dihedral angle of 6.61 (15)°. The thiazine ring adopts a sofa conformation. The toluene ring is oriented at dihedral angles of 15.5 (2) and 20.6 (2)° with respect to the bromobenzene and thiazine rings, respectively. The benzylidene system is approximately planar [r.m.s. deviation = 0.0388 Å]. In the crystal, weak intermolecular $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds connects the molecules into a chain along the b axis.

Related literature

For the synthesis of the title compound, see: Shafiq *et al.* (2011). For related structures, see: Khan *et al.* (2010); Shafiq *et al.* (2009); Arshad *et al.* (2009).



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Experimental

Crystal data

$\text{C}_{17}\text{H}_{16}\text{BrN}_3\text{O}_2\text{S}$
 $M_r = 406.30$
 Monoclinic, $P2_1$
 $a = 9.1077$ (6) Å
 $b = 6.8328$ (4) Å
 $c = 14.1765$ (9) Å
 $\beta = 96.807$ (3)°

$V = 876.00$ (10) Å³
 $Z = 2$
 Mo $K\alpha$ radiation
 $\mu = 2.48$ mm⁻¹
 $T = 296$ K
 $0.32 \times 0.12 \times 0.10$ mm

Data collection

Bruker Kappa APEXII CCD diffractometer
 Absorption correction: multi-scan (SADABS; Bruker, 2007)
 $T_{\min} = 0.504$, $T_{\max} = 0.790$

10166 measured reflections
 4119 independent reflections
 2881 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.026$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.034$
 $wR(F^2) = 0.078$
 $S = 0.97$
 4119 reflections
 220 parameters
 1 restraint

H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.29$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.36$ e Å⁻³
 Absolute structure: Flack (1983),
 1771 Friedel pairs
 Flack parameter: 0.004 (8)

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{C}17-\text{H}17\text{C}\cdots\text{O}1^i$	0.96	2.64	3.546 (5)	158

Symmetry code: (i) $-x + 2, y + \frac{1}{2}, -z$.

Data collection: APEX2 (Bruker, 2007); cell refinement: SAINT (Bruker, 2007); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: PLATON (Spek, 2009); software used to prepare material for publication: WinGX (Farrugia, 1999) and PLATON.

MS and MNA acknowledge the Higher Education commission of Pakistan for providing scholarships under its indigenous and IRSIP schemes.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: PV2429).

References

- Arshad, M. N., Zia-ur-Rehman, M. & Khan, I. U. (2009). *Acta Cryst.* **E65**, o3077.
 Bruker (2007). *SADABS, APEX2* and *SAINTE*. Bruker AXS Inc., Madison, Wisconsin, USA.
 Farrugia, L. J. (1999). *J. Appl. Cryst.* **32**, 837–838.
 Flack, H. D. (1983). *Acta Cryst.* **A39**, 876–881.
 Khan, I. U., Shafiq, M. & Arshad, M. N. (2010). *Acta Cryst.* **E66**, o2839.
 Shafiq, M., Tahir, M. N., Khan, I. U., Arshad, M. N. & Safdar, M. (2009). *Acta Cryst.* **E65**, o393.
 Shafiq, M., Zia-ur-Rehman, M., Khan, I. U., Arshad, M. N. & Khan, S. A. (2011). *J. Chil. Chem. Soc.* **56**, 527–531.
 Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
 Spek, A. L. (2009). *Acta Cryst.* **D65**, 148–155.